

## 2-Hydroxy-3-methoxybenzaldehyde (pyridinium-4-ylcarbonyl)hydrazone chloride hemihydrate

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## Key indicators

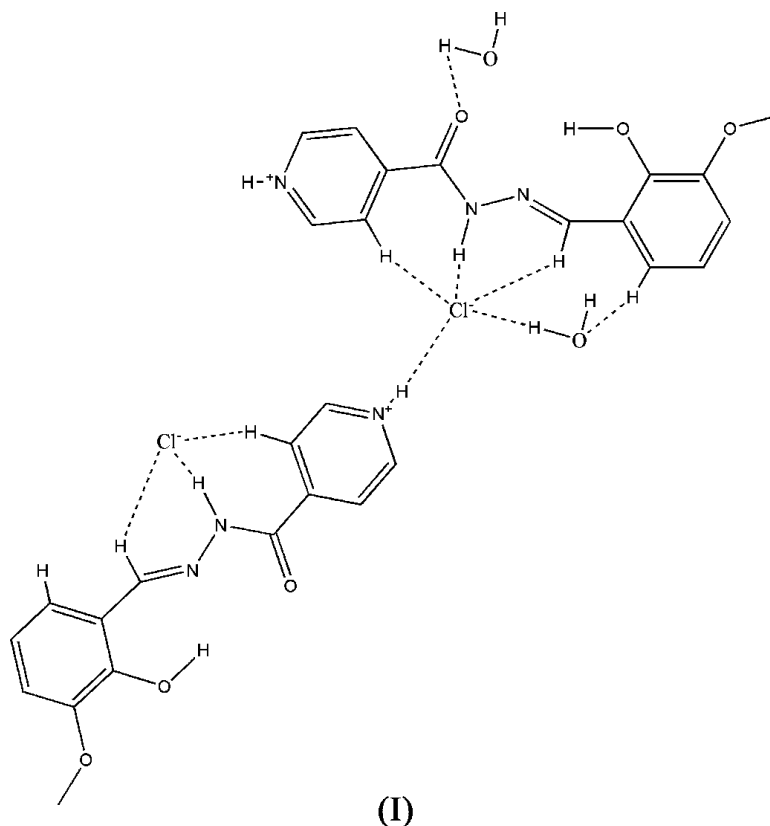
Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$   
Disorder in solvent or counterion  
 $R$  factor = 0.050  
 $wR$  factor = 0.146  
Data-to-parameter ratio = 13.0For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

The crystal structure of the title compound,  $\text{C}_{14}\text{H}_{14}\text{N}_3\text{O}_3^+\cdot\text{Cl}^-\cdot 0.5\text{H}_2\text{O}$ , exhibits  $\text{O}-\text{H}\cdots\text{O}$ ,  $\text{C}-\text{H}\cdots\text{O}$ ,  $\text{C}-\text{H}\cdots\text{Cl}$ ,  $\text{N}-\text{H}\cdots\text{Cl}$  and  $\text{O}-\text{H}\cdots\text{Cl}$  hydrogen bonds. The chloride anions participate in extensive hydrogen bonding with the aminium cations and link molecules through multiple  $\text{N}-\text{H}^+\cdots\text{Cl}^-$  interactions.

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## Comment

The molecular geometry of the title compound (Fig. 1) is listed in Table 1. The crystal packing (Table 2 and Fig. 2) is dominated by  $\text{N}-\text{H}^+\cdots\text{Cl}^-$ ,  $\text{C}-\text{H}\cdots\text{O}$ ,  $\text{C}-\text{H}\cdots\text{Cl}$  and  $\text{O}-\text{H}\cdots\text{Cl}$  hydrogen bonds, including also a three-centre  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bond (Jeffrey & Saenger, 1997). The intermolecular  $\text{N}-\text{H}\cdots\text{Cl}$ ,  $\text{C}-\text{H}\cdots\text{Cl}$  and  $\text{N}-\text{H}\cdots\text{Cl}$  bonds directly link two molecules (see scheme) through the chloride ion. The benzene rings are stacked along the  $c$ -axis direction by  $\pi-\pi$  interactions, forming a lipophilic layer, whereas hydrophilic layers are interconnected by  $\text{N}-\text{H}^+\cdots\text{Cl}^-$  hydrogen bonds.



Experimental

The title compound was prepared by the condensation of the *o*-vanillic and isonicotinic acid hydrazone (molar ratio 1:1) in ethanol/hydrochloric acid (3:1) at room temperature. The resulting solid was recrystallized from dichloromethane–ethanol (1:1, *v/v*). Yield 85%, m.p. 577–579 K. Analysis calculated for C<sub>14</sub>H<sub>15</sub>N<sub>3</sub>O<sub>3.5</sub>Cl: C 53.09, H 4.77, N 13.26%; found: C 52.98, H 4.69, N 23.15%.

Crystal data

C<sub>14</sub>H<sub>14</sub>N<sub>3</sub>O<sub>3</sub><sup>+</sup>·Cl<sup>-</sup>·0.5H<sub>2</sub>O  
*M<sub>r</sub>* = 316.74  
 Monoclinic, *P*2<sub>1</sub>/*n*  
*a* = 12.925 (3) Å  
*b* = 7.4369 (17) Å  
*c* = 15.503 (4) Å  
 β = 90.978 (4)°  
*V* = 1489.9 (6) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.412 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 μ = 0.27 mm<sup>-1</sup>  
*T* = 293 (2) K  
 Block, orange  
 0.17 × 0.15 × 0.10 mm

Data collection

Siemens SMART CCD area-detector diffractometer  
 φ and ω scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
*T<sub>min</sub>* = 0.955, *T<sub>max</sub>* = 0.973  
 7428 measured reflections  
 2580 independent reflections  
 1326 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.060  
*θ<sub>max</sub>* = 25.0°

Refinement

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.050  
*wR*(*F*<sup>2</sup>) = 0.146  
*S* = 1.00  
 2580 reflections  
 199 parameters  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.001P)^2 + 0.7491P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 (Δ/σ)<sub>max</sub> < 0.001  
 Δρ<sub>max</sub> = 0.25 e Å<sup>-3</sup>  
 Δρ<sub>min</sub> = -0.21 e Å<sup>-3</sup>

Table 1

Selected geometric parameters (Å, °).

N1–C1	1.346 (5)	N2–C7	1.280 (5)
N1–N2	1.370 (4)	O1–C1	1.223 (4)
C1–N1–N2	118.9 (3)	O1–C1–N1	122.4 (4)
C7–N2–N1	116.9 (3)	C3–C2–C1	124.1 (4)
C5–N3–C4	121.9 (4)		

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
N1–H1...Cl <sup>i</sup>	0.86	2.49	3.312 (3)	161
N3–H3...Cl <sup>ii</sup>	0.86	2.17	3.002 (3)	163
O2–H2...N2	0.82	1.91	2.632 (4)	146
O4–H15...Cl <sup>iii</sup>	0.85	2.33	3.146 (7)	161
O4–H16...O1	0.85	2.33	3.145 (7)	162
C5–H5...O4 <sup>iv</sup>	0.93	2.21	3.136 (8)	173
C7–H7...Cl <sup>i</sup>	0.93	2.71	3.549 (4)	151
C3–H3A...Cl <sup>i</sup>	0.93	2.74	3.662 (4)	172
C4–H4...O3 <sup>v</sup>	0.93	2.51	3.065 (5)	118
C4–H4...O2 <sup>v</sup>	0.93	2.33	3.248 (5)	170
C13–H13...O4 <sup>vi</sup>	0.93	2.57	3.237 (9)	129

Symmetry codes: (i)  $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $x, y + 1, z$ ; (iii)  $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$ ; (iv)  $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (v)  $x - \frac{1}{2}, -y + \frac{3}{2}, z - \frac{1}{2}$ ; (vi)  $-x + 1, -y + 1, -z + 1$ .

All H atoms were positioned geometrically and treated as riding on their parent atoms, with aromatic C–H distances of 0.93 Å and

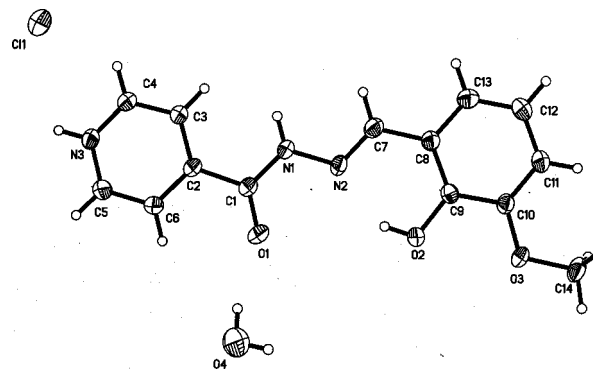


Figure 1 The structure of the title compound, with the atom numbering scheme and displacement ellipsoids shown at the 30% probability level.

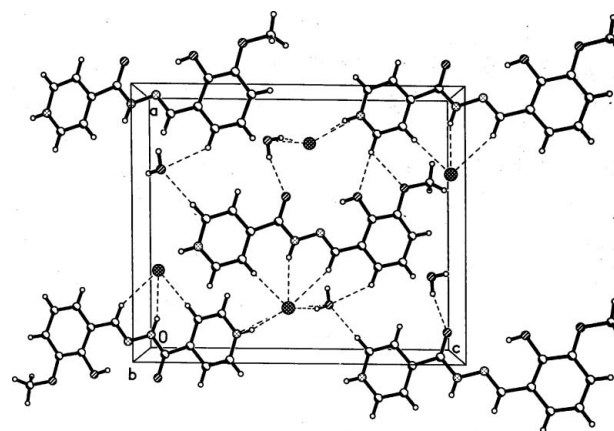


Figure 2 Crystal packing of the title compound. Dashed lines indicate hydrogen bonds.

methyl C–H distances of 0.96 Å (O,N–H as given in Table 2). The *U<sub>iso</sub>*(H) values were set at 1.2*U<sub>eq</sub>*(C,N) for the aromatic and N-bound H atoms, and 1.5*U<sub>eq</sub>*(C,O) for other H atoms.

Data collection: SMART (Siemens, 1996); cell refinement: SMART; data reduction: SAINT (Siemens, 1996); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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